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Class 12 Chemistry Chapter 2 Solutions The second chapter deals with types of solutions, and their properties. The concepts dealt with are Raoult's law, concentration of solutions, vapor pressure of liquid solutions, abnormal molar masses, and colligative properties.

NCERT Solutions for Class 12 Chemistry (Updated for 2020-21)

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Solutions . A solution is a homogeneous mixture of two or more components. The dissolving agent is the solvent. The substance that is dissolved is the solute. The components of a solution are atoms, ions, or molecules, making them 10^{-9} m or smaller in diameter.

Solutions, Suspensions, Colloids, and Dispersions

IDEAL SOLUTIONS; The solutions which obeys the Raoult's Law over the entire range of concentration are known as Ideal Solution. PROPERTIES : (i) $\Delta_{mix} H = 0$ i.e. The enthalpy of mixing of the pure component to form solution is 0. It is neither exothermic nor endothermic . (ii) $\Delta_{mix} V = 0$ i.e. There is no change in volume when 2 components are mixed to form solution.

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NCERT Solutions for Class 12 Chemistry Chapter 2 - Solutions. This article deals with NCERT Solutions for Class 12 Chemistry Chapter 2. Chemistry refers to the study of matter and its properties. This along with various formulae makes Chemistry a difficult subject. However, NCERT Solutions removes the fear of this subject.

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Solutions containing a precise mass of solute in a precise volume of solution are called stock (or standard) solutions. To prepare a standard solution, a piece of lab equipment called a volumetric flask should be used. These flasks range in size from 10 mL to 2000 mL and are carefully calibrated to a single volume.

13.7: Solution Dilution - Chemistry LibreTexts

b. No. of mole molecules = = = 2. Number of molecules = mole $\times 6.022 \times 10^{23}$. = $2 \times 6.022 \times 10^{23}$. c. c. Number of atoms = Number of molecules \times number of atoms in one molecules Number of atoms in one molecules of oxygen (O_2) = 2. \therefore total number of atoms = $2 \times 6.022 \times 10^{23}$

$23 \times 2. = 4 \times 6.022 \times 10^{23}$.

Kerala Syllabus 10th Standard Chemistry Solutions Chapter ...

The NCERT Solution Class 12 Chemistry Chapter 2 PDF covers the entire CBSE Chemistry syllabus and solutions to in text questions as well. Thus students are advised to follow these solutions and notes as not only they will help them to achieve more marks, but could also be used as the answer key to complete an assignment or homework faster.

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