

## Chemistry Ch 18 Solutions Practice Problems Answers

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### Chemistry Ch 18 Solutions Practice

360 Chemistry: Matter and Change • Chapter 18 Solutions Manual CHAPTER 18 SOLUTIONS MANUAL Section 18.3 Hydrogen Ions and pH pages 650–658 Practice Problems pages 651–657 22. The concentration of either the H<sup>1</sup> ion or the OH<sup>2</sup> ion is given for four aqueous solutions at 298 K. For each solution, calculate [H<sup>1</sup>] or [OH<sup>2</sup>]. State whether the solution is acidic,

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CHAPTER 18 Electrochemistry MULTIPLE CHOICE QUESTIONS NOTE: A table of standard reduction potentials is required to work many of these problems. 1. Complete and balance the following redox equation. When properly balanced using the smallest whole-number coefficients, the coefficient of S is H<sub>2</sub>S + HNO<sub>3</sub> → S + NO (acidic solution) 2.

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